

| Course Type | Course Code | Name of Course         | L | T | P | Credit |
|-------------|-------------|------------------------|---|---|---|--------|
| DC          | NCHC508     | Advanced Mass Transfer | 3 | 1 | 0 | 4      |

#### Course Objective

- To extend the fundamental concepts of mass transfer in cases of multi-component systems (with and without chemical reactions) and the application of those concepts in solving real engineering problems.

#### Learning Outcomes

- Students shall have adequate knowledge in tackling engineering problems with complex mass transfer operations.

| Unit No. | Description of Lectures   | Class Hours | Learning Outcomes  |
|----------|---|-------------|--|
| 1.       | <b>Review:</b> VLE and VLLE of binary system, Introduction to Multi-phase equilibrium, ternary diagram, residual curve, ideal solution of multi-components  | 3 L + 1 T   | Student will learn the basics of vapor-liquid equilibrium.   |
| 2.       | <b>Thermodynamic correlation:</b> Multi-phase equilibrium, K-value and activity coefficient   | 3 L + 1 T   | Student will learn the basic of thermodynamic relations.   |
| 3.       | <b>Multi-component diffusion and convection:</b> Multi-component diffusion, convective mass transfer, correlations for mass transfer coefficients, review of the models for mass transfer at fluid-fluid interface. | 6 L + 2 T   | Student will learn fundamental concepts of diffusion and convection especially for multi-component system. |
| 4.       | <b>MESH equations:</b> Formulation of Material and Energy balance equation (MESH) for single stage operation, batch distillation, flash vaporization, degrees of freedom (DOF)                                      | 6 L + 2 T   | Student will learn mathematical concept of single stage mass transfer operation.                           |
| 5.       | <b>Short-cut methods:</b> Short-cut methods for designing multi-component multi-stage fractionation: Kremser equation, Fenske-Underwood-Gilliland, Wang-Henke, Naphtali-Sandholm, Thiele-Geddes                     | 3 L + 1 T   | Student will learn the different easy mathematical method for designing of distillation unit.              |
| 6.       | <b>Rigorous methods:</b> Rigorous method of multi-component multi-stage fractionation: MESH Equations for multi-stage fractionation, DOF, Solution technique  | 6 L + 2 T   | Student will learn the rigorous method for designing of distillation unit.                                 |
| 7.       | <b>Multicomponent distillation:</b> Multi-component distillation and cascading of columns, divided-wall distillation columns  | 6 L + 2 T   | Student will learn operation techniques of distillation unit for multi-component system.                   |

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|----|--|-----------|---|
| 8. | <b>Reactive mass transfer:</b> Introduction to reactive distillation (RD) process, advantages and disadvantages of RD, design of RD columns. Mass transfer: Gas liquid reactions: solutions for slow, fast and instantaneous reactions with adsorption for single and two gases. | 6 L + 2 T | Student will learn the reactive distillation process.         |
| 9. | <b>Supercritical Fluid Extraction:</b> Supercritical fluid extraction  | 3 L + 1 T | Student will learn the supercritical fluid extract phenomena. |
|    | <b>Total</b>   | <b>56</b> |   |

#### **Textbooks:**

1. Treybal, R.E. (1981). Mass Transfer operations, 3<sup>rd</sup> Ed. McGraw Hill Publication.
2. Seader, J. D., Henley, E. J., & Roper, D. K. (1998). Separation Process Principles. John Wiley & Sons.

#### **Reference Books:**

1. Taylor, R., & Krishna, R. (1993). Multicomponent mass transfer. John Wiley & Sons.
2. Doraiswamy, L.K. & Sharma, M.M (1984) Heterogeneous reaction: analysis, examples and reactor design vol2 , Fluid-fluid Solid reaction, Wiley.
3. Kulprathipanja, S. (2002). Reactive separation processes. Taylor & Francis: New York.